

Huges s Science Readers



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Reviews

This written book is excellent. It really is rally fascinating throgh studying period. You are going to like the way the writer write this publication.

(Hadley Ullrich)

HUGES S SCIENCE READERS



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1884 Excerpt: .with one end in the fire, very soon the other end will become too hot for you to maintain your hold. Of this you have a familiar example when a poker is put in the fire. But if you hold one end of a long stick of charcoal in a flame, the charcoal will burn at that end, but you will have no difficulty in maintaining your hold of the other end. This difference is expressed by saying that metals are better conductors of heat than non-metals. Of course the metals are not all exactly alike in other particulars, though they all agree in having a bright surface when polished, and in being good conductors of heat. Metals differ much in colour as well as in other respects; copper is red, silver is white, gold yellow, iron grey, and so on. One or two elements, like arsenic, in some particulars resemble metals, but in others resemble non-metals, and it is doubtful under which head we ought to put them. Here is a list of some of the commoner elements, classified into metals and non-metals: --Metals. Non-Metals. Mercury Oxygen Silver Hydrogen Gold Nitrogen Iron Carbon Lead Chlorine Zinc Sulphur Nickel Silicon Sodium Phosphorus Potassium Two or more elements combine together to form compounds. But we may have two or more elements mechanically mixed together, and yet not forming a compound. What then constitutes the difference between a mechanical mixture and a chemical compound? Perhaps we can best...



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