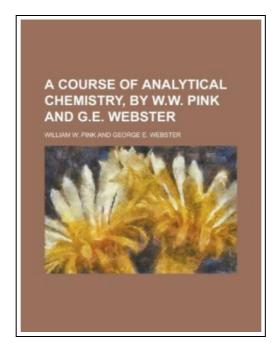
A Course of Analytical Chemistry, by W.W. Pink and G.E. Webster



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A COURSE OF ANALYTICAL CHEMISTRY, BY W.W. PINK AND G.E. WEBSTER



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Rarebooksclub.com, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****. This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1874 edition. Excerpt: . Acid. 121. Add MnO2, manganic oxide (black oxide of manganese), and SO, Ho2, then apply heat; evolution of chlorine takes place, known by its smell, colour, c. 122. NO2Ago gives a curdy-looking white ppt. of AgCl, argentic chloride (chloride of silver), soluble in AmHo, but insoluble in weak NO2Ho; exposure to the sunlight always darkens this ppt. HBr.--Hydrobromic Acid. 123. Add MnO2 and SO2Ho2, then apply heat; evolution of bromine vapour takes place, which colours moistened starch paper ytlhw. 124. NO2Ago gives a cream-coloured yellow ppt. of AgBr, sightly soluble in AmHo, but insoluble in N02Ho. HI.--Hydriodic Acid. 125. Treat as in Reaction No. 123; evolution of iodine vapour of a violet or purple colour takes place, which colour moistened starch paper blue. 126. NO2Ago gives a light yellow ppt. of Agl, argentic iodide (iodide of silver), soluble in KCy, but insoluble in weak NO2Ho. No 127. ij. Hgo, mercuric nitrate (nitrate of mercury), gives a greenish-yellow ppt., insoluble in excess of the reagent. SH2.--Hydrosulphuric Acid. 128. Addition of HCl and application of heat causes evolution of SH2, known by its nasty smell; sometimes a piece of zinc is required in the case of those sulphides insoluble in HCl. C(Me)O 129. Pbo gives a black ppt. of Pl)S, plumbic C(Me)O sulphide (sulphide of lead), with soluble sulphides. coiio AcEXIC AciD (This acid is only partly precipitated by NO2Ago.) 129a. Add Fe2Cl6, ferric chloride (perchloride of iron), to acetic solutions; it will change them...



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