## Chemistry of Common Objects Adapted to the Alternative Elementary Stage of the Syllabus of the Department of Science and Art



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### Reviews

Very helpful to all of class of folks. This is certainly for all who statte there had not been a worthy of studying. Once you begin to read the book, it is extremely difficult to leave it before concluding.

(Jayda Lehner Jr.)

# CHEMISTRY OF COMMON OBJECTS ADAPTED TO THE ALTERNATIVE ELEMENTARY STAGE OF THE SYLLABUS OF THE DEPARTMENT OF SCIENCE AND ART



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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1892 Excerpt: that the carbon dioxide has been absorbed. We leant from this that alkalies will absorb carbon dioxide, and, on the other hand, we have seen in the last lesson that the common acids will liberate carbon dioxide from sodic carbonate. Experiment 138.--Into an evaporating dish pour about 25 cc. of dilute sulphuric acid, colour the same with litmus solution. Prepare a strong solution of potash, and add this to the acid, drop by drop, with constant stirring, until the liquid is just neutralised, as in experiment 134. Evaporate one portion of the liquid nearly to dryness, and then allow it to cool. Tour out another portion into a dinner plate. Observe that the sulphuric acid is neutralised by the potash, and that a Btage is reached as the alkali is added to the acid, when at last a liquid is obtained that has no action on litmus. On heating and evaporating oil some of the water a new body is obtained in the form of crystals; these have a decided saline taste We learn from this, once again, that by the combination of an alkali and an acid, here potash and sulphuric acid, a salt is produced; in this case sulphate of potash, or potassium sulphate. In a similar manner, by neutralisingportions of sulphuric acid with soda or ammonia, the salts, sulphate of soda and sulphate of ammonia, may be prepared. SALTS are bodies which are formed when acids are partly or entirely neutralised by a...

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