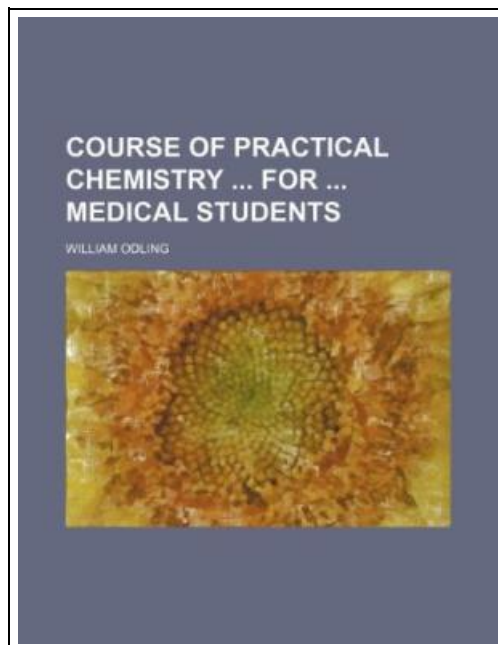


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book \*\*\*\*\* Print on Demand \*\*\*\*\*.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1865 Excerpt: . sulphites, there is produced a dark reddish-brown colouration, which disappears on the addition of hydrochloric acid, or on boiling: in the latter case with formation of a red-brown turbidity. Nitrates and chlorates are unaffected by any of the general reagents mentioned in the table. (80.) Although nearly all of the acids react more or less characteristically with several reagents as above described, the following list of acids or salts, with the tests by which their presence is more particularly indicated or established, may nevertheless prove useful to the student: --V Deflagration on charcoal, and reaction with sulphuric acid. Chromates Reaction with sulphydric acid. Nitrates Chlorates Carbonates Sulphides! Effervescence with acids. Sulphites / Sulphates Precipitation by chloride of barium. In the following paragraphs, describing the several reactions of the different acids, the order of the above list is adopted. (81.) Chromates. Chromates are usually of a yellow or red colour, and, except those of the alkali metals, are more or less insoluble in water. a. Chromates acidified with hydrochloric acid, and treated with sulphuretted hydrogen, yield a deposit of sulphur, and a green solution of chromic chloride  $\text{Cr}_2\text{Cl}_3$ . The same compound is produced on boiling the solution of a chromate with hydrochloric acid and a little alcohol. To it the ordinary tests for chrome salts can be applied. (5. Nitrate of silver gives a dark red precipitate of chromate of silver  $\text{Ag}_2\text{CrO}_4$ , soluble in nitric acid. Nitrate of barium and acetate of lead throw down yellow...



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